## THERMODYNAMICS-I 4<sup>th</sup> Exam/Mech./5253/Jun'2022 (For 2018 Batch Onwards)

## Duration: 3Hrs.

#### SECTION-A

M.Marks:75

## Q1. Do as directed.

15x1=15

- a. Solids are heated by Conduction.
- b. Heat and Work are mutually\_\_\_\_
- c. Free expansion is a \_\_\_\_\_Process.
- d. Locomotive boiler is a \_\_\_\_\_tube boiler.
- e. Otto cycle is also known as \_\_\_\_
- f. The efficiency of Carnot cycle is maximum.
- g. An \_\_\_\_\_\_Is a device which compresses the air and raises its temperature?
- h. The Mollier diagram is a plot of enthalpy and entropy.
- i. Dryness fraction of dry steam is \_\_\_\_\_
- j. For adiabatic process heat is \_\_\_\_\_
- k. Entropy of the universe remains constant.
- I. Work is a point function.
- m. Intensive property is independent upon mass.
- n. The units of Thermal conductivity is \_\_\_
- o. The value of Cv for a gas is always less than Cp.

#### **SECTION-B**

## Q2. Attempt any six questions.

- i. Define thermodynamic system. Explain their types briefly.
- ii. State Charle's law.
- iii. Define Isothermal process. Give its relation for heat and work.
- iv. Derive ideal gas equation.
- v. State first law of thermodynamics.
- vi. What do you mean by Mollier diagram?
- vii. Define Fourier's law.
- viii. State Zeroth law of thermodynamics.
- ix. State the function of boiler mountings.

### SECTION-C

# Q3. Attempt any two questions.

- a. Explain joule's experiment for first law of thermodynamics.
- b. State and explain Kelvin Planck statement and Clausius statement for 2<sup>nd</sup> law of thermodynamics.
- c. Explain with neat sketch working of Babcock and Wilcox boiler.
- d. i) List the use of compressed air in human life.ii) Define Entropy and give its formula.iii) Define Conduction and convection.
- 2x15=30

6x5=30